



**Developing and Characterization of Nonionic Microemulsions
Containing Ferrous Sulfate and Folic Acid for Transdermal
Application**

By

Hussain Yousef ALGhatm

Supervised by

Dr. Jamal Alyoussef Alkrad

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Faculty of Pharmacy

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COMMITTEE DECISION

Thesis (Developing and Characterization of Nonionic Microemulsions
Containing Ferrous Sulfate and Folic Acid for Transdermal Application)

Examination Committee

Signature

Dr. Jamal Alyoussef Alkrad (Supervisor)

Assist.Prof of Pharmaceutical Technology and Biopharmacy

Dr. Ahmed A. Talhoni (Examiner)

Assist.Prof. of Pharmacology

Dr. Hatim Alkhatib (Examiner)

Associate professor of industrial pharmacy and pharmaceuticals

AUTHORIZATION STATEMENT

I'M HUSSAIN YOUSEF ALGHATM, Authorize Isra University to supply hard and electronic copies of my thesis to libraries, establishments, or bodies and institutions concerned with research and scientific studies upon request, according to the university regulations.

Name: Hussain Yousef Alghatm

Date: 7 May 2018

Signature:

Dedication

To my mother (Safia), who leads me through the valley of darkness

with light of hope and support.

To my father (Yousef), who stands by me when the things look gloomy.

To my family, the symbol of love and giving.

To anyone who has shown me friendship and Kindness during my life.

*Whose effective, Love, encouragement and prays of day and night make
me able to get success.*

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LIST OF ABBREVIATION AND SYMBOLS

SAMBOL ABBREVIATION	DEFINITION
DMSO	Dimethyl Sulfoxide
Span 20	Sorbitan mono laurate
Tween 80	Poloxyethylene Sorbitan mono-oleate
Vol.	Volume
IPM	Isopropyl Myristate
W/O	Water in Oil
W	Water
O/W	Oil in Water
μEm	Microemulsion
HLPC	High Pressure Liquid Chromatography
GI	Gastrointestinal
SC	Stratum corneum
TDDS	Transdermal drug delivery system
FPP	Ferric Pyrophosphate
IDA	Iron Deficiency Anemia
FeSO ₄	Ferrous Sulfate
PDI	Polydispersity index
FTIR	Fourier transform infrared spectroscopy

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Abstract

Ferrous sulfate and Folic acid have been widely used for the treatment of iron deficiency such as in anemia or pregnancy. Because they are ingested by oral they are leading to many of complication, for instance, GI disturbances (constipation, nausea, and unpleasant taste), and dangerously high blood pressure, severe headache, blurred vision and confusion with anxiety if taken as parenteral route. In the other hand, they are also interacting with numerous drugs, food and minerals). In this study, four FeSO₄ μEm and 8 folic acid μEm systems were developed for transdermal application purpose to overcoming these problems resulted from Oral and parenteral routes. These W/O μEm are prepared by nonionic surfactants (Span20 & Tween80), IPM as oil phase, and water as aqueous phase. The μEm for both FeSO₄ and Folic acid were characterized for their droplet sizes, zeta potential, pseudo-ternary phase diagram, and rheological properties. Moreover, the FTIR used to prove the encapsulation of folic acid in the μEm. The investigation demonstrate that FeSO₄ and Folic acid μEm are correspond to the characteristics and properties of the μEm in literature. In this study, although, the most Microemulsions prepared have ideal characters to the Microemulsions properties written in literature. The results are shows that a significant variance between droplet size, Zeta

potential, rheological properties and characterization of Microemulsions. The surfactants used in this study shows a factor that may play a role in preparation of those Microemulsions, and also the ratio of these surfactants have an important function to produce an ideal Microemulsion.